



Centre for Energy Research

Utilization of nuclear power plants for
hydrogen production

Attila R. Imre

(imre.attila@ek-cer.hu)



Mature	Promising	Lab-scale
Electrolysis (alkaline) of <i>water</i>	Electrolysis (SOE) of <i>water</i>	Other biological (like fermentation) from <i>biomass</i>
Electrolysis (PEM) of <i>water</i>	Dry <i>methane</i> reforming/pyrolysis	Radiolytic decomposition of <i>water</i>
Steam <i>methane</i> reforming	Thermolysis of <i>water</i> (direct thermal decomposition)	Microbial electrolysis of fermented <i>biomass</i>
	Thermochemical decomposition of <i>water</i>	Artificial photosynthesis (photocatalytic process) of <i>water</i>
	Thermochemical decomposition of <i>organic material (fossile)</i>	
	Partial oxidation of <i>methane</i> (SynGas)	
	<i>Coal</i> gasification (with <i>water</i>)	
	<i>Biomass</i> gasification	



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Good hydrogen:

- From water or from renewable organic materials
- Zero or nearly zero carbon emission during the production



Three “products” of a nuclear power plant, which can be utilized for hydrogen production:

- Electricity
- Heat
- Radiation – probably not relevant, omitted here.



Utilization of heat:

- For thermolysis: at least 800-1200 °C, does not work with traditional reactors (water used as moderator, cannot reach degradation temperature!). Concerning novel reactors (molten salt/metal or He-cooled ones) the temperature can be reached, but transportation can be very difficult (long/safe distance transport, loss of heat, extensive amount of heat transfer material) .



Utilization of heat:

- For thermochemistry: special chemical reactions with multi-step water cracking. Temperature is slightly lower than for direct pyrolysis, but still quite high, 600-900 °C; might work for GEN-IV. Potential co-generation (heat for hydrogen production, electricity to grid). Extensive research for reactions running on lower temperature!



Utilization of heat:

- Biological/biochemical processes for hydrogen production, needs very low temperature (sometimes even tertiary waste heat can be utilized), but presently the efficiency is low. Only in laboratory-scale, probably far from maturity. Extensive research on the increase of efficiency...might be the way of future.



Utilization of heat:

- Problem: in heat-based methods, usually hydrogen and oxygen are generated in the same location; explosive mixture, not favourable around nuclear installations.





Utilization of electricity:

Electricity from NPPs is definitely carbon-free, more or less green (at least in the next few decades), although definitely not renewable. NPPs are baseload-plants, providing nearly constant power (hence the hydrogen-production would be predictable, unlike for most renewables, except geothermal-based one, where the capacity is usually too small).



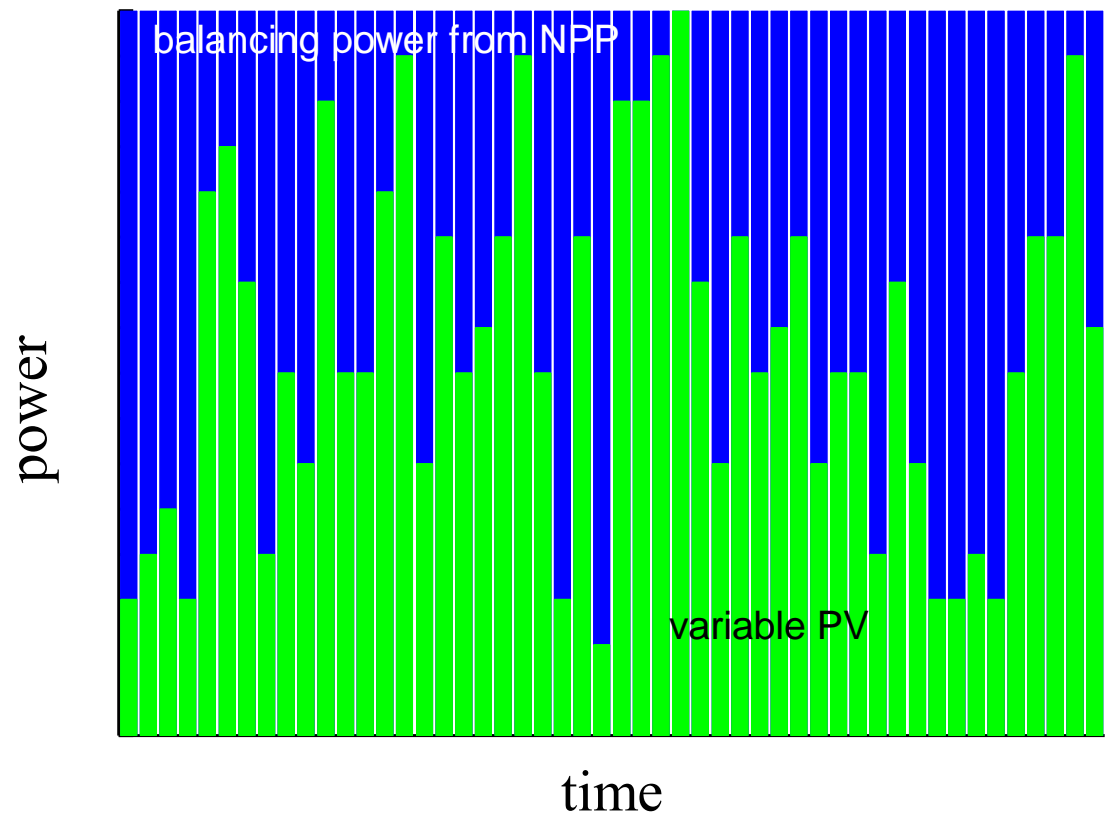
PEM electrolyzers can handle fluctuating power (like PV-based), while alkaline (non-expensive) or SOE (very efficient) electrolyzers need constant power, hardly possible to provide with wind+solar.

- One can build huge PV-parks, where the “base” part of the power plus a small amount (smoothened by batteries) can go to SOE to produce a lot of hydrogen or
- One can use an existing or new NPP for the same purpose.

Infrastructure (for hydrogen transport) is usually better around NPP-s; costs, and public opinion is probably against them.



Potential solution: NPP + off-grid high-capacity PV (in the same location, i.e. only local grid needed):



Constant power, even alkaline electrolyzers (accessible!). Existing NPP-s with nearby PV-fields can be utilized immediately; for NPP-s without PV, the solar part can be built in a few months.





What is this? a Nuclear Power Plant???



Temelin



Presently for the production of industrial-quantity hydrogen with NPP-s, the NPP-levelled PV-based (partly renewable power) mass-electrolysis (probably alkaline- or SOE-type) can be a promising solution!



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Thank you for your
attention!